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Multiple Choice Quiz - Electronic Structure of Atoms I

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(a) Nucleus of an atom contains no negative charge (b) Electrons move in circular orbits around the nucleus (c) Electrons occupy orbitals of lowest energy (d) All the four

Electron Configurations and the Periodic Table Multiple Choice

Quantum Mechanics Question: That radiation and matter have properties both of particles and of waves is called what?

Answer: Louis de Broglie suggested in 1924 that matter, like radiation, had both particle and wave

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Which one of the following electrons, identified only by their n and l quantum numbers, has the lowest energy? n = 3, l = 1 n = 3, l = 0 n = 2, l = 1 n = 3, l = 2 A) n = 2, l = 1: B) n = 3,

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1 Answer Key Testname:

QUIZ_QUANTUM_THEORY_CH_07 TST MULTIPLE CHOICE Choose the one alternative that best completes the statement or answers the question

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Class 11 Chemistry MCQ - Quantum Mechanical Model of Atom

1 Which one of the following sets of quantum numbers is valid? 1 n l ml s A 2 3 -3 +1/2 B 4 3 4 -1/2 C 3 1 0 +1/2 D 1 -1 1 +1/2 E 2 1 2 -1/2 2 How many protons (p), neutrons (n)

Quantum Numbers MCQs for NEET with Answers - BYJU'S

Chemistry Multiple Choice Questions on "Quantum Mechanical Model of Atom" 1 The principal quantum number describes ___ a) energy and size of the orbit b) the shape of

The University of Sydney

1 Which of the following quantum numbers can distinguish between two electrons present in the same orbital? (a) Azimuthal quantum number (b) Principal quantum number (c)

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Q1 Choose the wrong statement about the spin of an electron, according to quantum mechanics: a) It is related to intrinsic angular momentum b) Spin is the rotation of an

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1) Which of the following statements is correct for an electron that has the quantum numbers n = 4 and ml = -2? A the electron may be in a p orbital B the electron may be

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