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C H O O O CO CO CO H O H O H O number of moles of unknown substance O CO H O C H CO H O C H O H O C H O CO use mole ratio 5 mol O 2 1 mol C 3 H 8 3 mol CO 2 1 mol C 3 H 8 4 mol H 2 O 1 mol ...MATH SKILLS TRANSPARENCY MASTER 16 Name Date Class MATH SKILLS TRANSPARENCY MASTER 4 ... 94 Chemistry: Matter and Change • Chapter 4 Math Skills Transparency Masters Calculating Atomic Mass MATH SKILLS TRANSPARENCY MASTER Use with Chapter 4,

Section 4.3 4 Given the data in the table, calculate the atomic mass of unknown element X. Then, identify the unknown element, which is used Math Skills Transparency Master Chapter 13 1 MATH SKILLS TRANSPARENCY MASTER. Use with Chapter 4, Section 4.3. 4. Given the data in the table, calculate the atomic mass of unknown element X. Then, identify the unknown element, which is used medically to treat some mental disorders. Analyze the Problem. You

are given the data in the table. Name Date Class MATH SKILLS TRANSPARENCY MASTER 4 ...Chapter 11, 1.2 MATH SKILLS TRANSPARENCY WORKSHEET 1. Explain what a crystalline solid is. 18 Use with Chapter 12, Section 2.3 1. The reaction of iron(III) oxide ( $\text{Fe}_2\text{O}_3$ ) and hydrogen ( $\text{H}_2$  is represented by the following unbalanced chemical equation.  $\text{Fe}_2\text{O}_3(s) + \text{H}_2(g) \rightarrow \text{Fe}(s) + \text{H}_2\text{O}(l)$  MATH SKILLS TRANSPARENCY MASTER 16 Teaching Transparency Worksheet Chapter

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 Numbers of Ions Use with  
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 Write a simple word  
 equation that illustrates  
 what must be true of total  
 positive charge and total  
 negative charge in an  
 ionic compound. 2. How  
 many potassium ions  
 (group 1) would be  
 related to balance the  
 charge of each of the  
 following in a  
 compound?Snow  
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 Worksheets 1. In equation  
 1, what do the numbers  
 238 and 92, written to the  
 left of the symbol U,  
 represent? 2. Explain how

both mass number and atomic number have been conserved in equation 1.

3. Write the nuclear symbol for the missing term in equation 2.

4. balancing nuclear equations

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Look at the two waves shown. What is the speed of each wave?

2. Look at the two waves shown. Which wave has a higher frequency? Which wave has a longer wavelength?

3. Assume that wave A has a wavelength of 699

nm. Calculate the frequency of the wave. Show your work.

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both mass number and atomic number have been conserved in equation

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Numbers of Ions Use with

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Write a simple word

equation that illustrates

what must be true of total

positive charge and total

negative charge in an

ionic compound 2 How

many potassium ions

(group 1)

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WORKSHEET Determining

Numbers of Ions Use with

Chapter 7, Section 7.3 1.

Write a simple word

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(group 1) would be

related to balance the

charge of each of the

following in a compound?

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known substance C H C H

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H O H O number of moles

of unknown substance O

CO H O C H CO H O C H O

H O C H O CO use mole

ratio 5 mol O 2 1 mol C 3

H 8 3 mol CO 2 1 mol C 3

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Numbers of Ions Use with

Chapter 7, Section 73 1  
Write a simple word equation that illustrates what must be true of total positive charge and total negative charge in an ionic compound 2  
How many potassium ions (group 1)

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