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LILLY CARLA

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 I Chemistry 1 What is the difference between a homogeneous mixture and a heterogeneous mixture? 2 List 2 examples of a homogeneous mixture and 2 examples of heterogenous mixtures 3 List 2 examples of a pure substance and 2 examples of a mixture 4

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Physical Science: High School Final Exam Free Practice Test Instructions: Choose your answer to the question and click 'Continue' to see how you did Then click 'Next Question' to answer the **PHYSICAL SCIENCE FINAL EXAM** chemistry Which of the following is a positive test for the presence of nickel?

(a) a characteristic pink color in the presence of dimethylglyoxime (b) skin allergies when jewelry contains nickel (c) a bright yellow precipitate in the presence of chloride ions (d) a strong odor in the presence of an acid Verified answer

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Physical Science Final Exam: Table of Contents I Metric Prefixes, Defined Units, Derived Units, Conversions, Addition and Subtraction with Units, Multiplication with Units II Scientific Measurement: Addition and Subtraction with Units, Multiplication and Division with Units , Accuracy, Precision, Percent Error, Percent Difference III

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PHYSICAL SCIENCE FINAL EXAM STUDY GUIDE - FIRST SEMESTER What instrument is the best to use if you want to measure a liquid? What are the steps of the scientific method and what approximate order do they occur? Which type of graph is best when showing changes over a period of time?

Physical Science Final Exam Review The Atom Protons (p+) positive in charge located in nucleus gives identity to element Neutrons (n0) Neutral located in nucleus gives mass to element largest mass Electrons (e-) a Negatively charged b located in energy levels in electron cloud outside the nucleus